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# Metals & Traces®

## **HEAVY METALS & TRACE MINERALS EXTENDED PROFILE, WHOLE BLOOD**

Name:	Gender:	Age:
Sample Taken Day:	Date of Analysis:	
Specimen: Whole Blood	Patient Code Number:	

Ess	Essential Trace Elements					
Code	TEST		RESULT		REFERENCE VALUES	
344	Vanadium ( <b>V</b> ), Whole Blood	23 <b>V</b> Vanadium 50.942	0.125 μg/L	•	< 0.800 μg/L	
1118	Iodine (I), Whole Blood	53     lodine   126.904	48.486 μg/L	·	15.000 – 132.000 μg/L	
1203	Cobalt ( <b>Co</b> ), Whole Blood	27 <b>Co</b> Cobalt 58.933	0.187 μg/L		< 1.500 μg/L	
1309	Manganese ( <b>Mn</b> ), Whole Blood	25 Mn Manganese 54.938	3.64 <b>7</b> μg/L	$\downarrow$	7.100 – 20.000 μg/L	
1936	Magnesium ( <b>Mg</b> ), Whole Blood	12 Mg Magnesium 24.305	21.902 mg/L	<b>\</b>	Adults 30.000 – 55.000 mg/L Children 25.000 – 48.500 mg/L	
1348	Molybdenum ( <b>Mo</b> ), Whole Blood	42 <b>Mo</b> Molybdenun 95.95	<b>0.250</b> μg/L	<b>\</b>	Adults 0.300 – 1.800 μg/L Children 0.500 – 1.800 μg/L	
1937	Selenium ( <b>Se</b> ), Whole Blood	34 Se Selenium 78.972	<b>142.655</b> μg/L	1	Adults 79.00 – 130.00 μg/L Children 60.00 – 120.00 μg/L New reference range from 1/6/2024	
1938	Copper ( <b>Cu</b> ), Whole Blood	29 <b>Cu</b> Copper 63.546	0.597 mg/L	<b>\</b>	Adults 0.756 – 1.500 mg/L Children 0.600 – 1.360 mg/L New reference range from 1/6/2024	

Reference Values & Methods adapted from:

1. Analytical Biochemistry, Holme & Peck, 3<sup>rd</sup> ed., 1998, Prentice Hall

2. Laboratory Tests and Diagnostic Procedures, Chernecky & Berger, 5<sup>th</sup> ed., 2008, Saunders Elsevier 3. Interpretation of Diagnostic Tests, Wallach, 8<sup>th</sup> ed., 2007, Lippincott

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Specimen: Whole Blood	Patient Code Number:	

Ess	Essential Trace Elements					
Code	TEST		RESULT		REFERENCE VALUES	
1845	Chromium ( <b>Cr</b> ), Whole Blood	24 Cr Chromium 51.996	0.808 μg/L	•	Adults < 2.000 μg/L Children < 1.000 μg/L	
1939	Zinc ( <b>Zn</b> ), Whole Blood	30 <b>Zn</b> Zinc 65.38	3.908 mg/L	<b>\</b>	4.000 – 7.500 mg/L	

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Name:	Gender:	Age:
Sample Taken Day:	Date of Analysis:	
Specimen: Whole Blood	Patient Code Number:	

Pot	Potentially Toxic Elements					
Code	TEST		RESULT		REFERENCE VALUES	
146	Aluminum ( <b>Al</b> ), Whole Blood	13 Al Aluminum 26.982	0.216 μg/L	•	<30.000 μg/L	
224	Antimony ( <b>Sb</b> ), Whole Blood	51 <b>Sb</b> Antimony 121.760	8.441 μg/L	<b>↑</b>	< 3.500 μg/L	
311	Silver ( <b>Ag</b> ), Whole Blood	47 <b>Ag</b> Silver 107.868	0.200 μg/L		Adults < 2.000 μg/L Children < 0.600 μg/L	
319	Arsenic ( <b>As</b> )-Toal, Whole Blood	33 <b>As</b> Arsenic 74.922	1.960 μg/L	•	< 10.000 μg/L	
376	Beryllium ( <b>Be</b> ), Whole Blood	4 Be Beryllium 9.012	0.103 μg/L	•	< 0.400 μg/L	
387	Bismuth ( <b>Bi</b> ), Whole Blood	83 Bi Bismuth 208.980	0.146 μg/L	•	< 1.000 μg/L	
990	Zirconium ( <b>Zr</b> ), Whole Blood	<sup>40</sup> <b>Zr</b> Zirconium 91.224	0.149 μg/L	•	< 3.000 μg/L	
1027	Thallium (TI), Whole Blood	81 <b>TI</b> Thallium 204.383	0.125 μg/L	•	< 0.600 μg/L	
1125	Cadmium ( <b>Cd</b> ), Whole Blood	48 <b>Cd</b> Cadmium 112.411	0.500 μg/L	•	Adults < 1.100 μg/L Children < 0.500 μg/L	

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Name:	Gender:	Age:
Sample Taken Day:	Date of Analysis:	
Specimen: Whole Blood	Patient Code Number:	

Pot	Potentially Toxic Elements				
Code	TEST		RESULT		REFERENCE VALUES
1181	Tin ( <b>Sn</b> ), Whole Blood	50 <b>Sn</b> Tin 118.711	0.500 μg/L	•	< 1.300 μg/L
1278	Platinum ( <b>Pt</b> ), Whole Blood	Platinum 106.42	0.250 μg/L	•	Adults < 0.400 μg/L Children < 0.500 μg/L
1355	Lead ( <b>Pb</b> ), Whole Blood	82 <b>Pb</b> Lead 207.2	5.411 μg/L		Adults < 30.000 μg/L Children < 35.000 μg/L
1412	Nickel ( <b>Ni</b> ), Whole Blood	28 <b>Ni</b> Nickel 58.693	0.204 μg/L	•	< 2.000 μg/L
1473	Uranium ( <b>U</b> ), Whole Blood	92 <b>U</b> Uranium 238.029	0.075 μg/L	•	< 0.100 μg/L
1748	Mercury ( <b>Hg),</b> Whole Blood	80 <b>Hg</b> Mercury 200.952	1.431 μg/L	•	Adults < 2.000 μg/L Children < 1.000 μg/L

\* ND = Not Detected



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### Metals & Traces®

### **HEAVY METALS & TRACE MINERALS EXTENDED PROFILE, WHOLE BLOOD**

Name:	Gender:	Age:
Sample Taken Day:	Date of Analysis:	
Specimen: Whole Blood	Patient Code Number:	

#### **General Notes and Comments**

The information contained in this report is an interpretive aid to diagnostic procedures. The findings should be related to clinical examination, individual medical history, and possibly other diagnostic tests. Reference values have been obtained from the CDC (Center for Disease Control and Prevention, USA), WHO (World Health Organization), and other international agencies.

Blood is a transport medium, and the concentrations of metals and trace elements found in the blood reflect the immediate nutritional status of the body as well as the factors affecting their uptake and distribution. The presence of toxic metals in the blood is often associated with direct exposure and indicates the need for medical attention. Reduced concentrations of micronutrients and trace elements reflect insufficient intake and may indicate nutritional deficiency. The minerals circulate in the blood for about 72 hours and are either excreted or deposited in various body tissues.

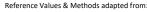
Determining heavy metals in the blood is the most reliable test for diagnosing toxicity or poisoning from metals such as lead, mercury, etc. The Heavy Metals and Trace Minerals in Whole Blood reflect the levels of elements regardless of their form (organic or inorganic), both intracellular (within blood cells) and extracellular (in serum or plasma) in total (without differentiation).

Heavy metals become toxic to organisms because they can replace other elements in the macromolecules of cells (e.g., enzymes, proteins, etc.). For example, Cadmium (Cd) toxicity occurs because it can replace Zinc (Zn) in many essential enzymes that contain Zinc in their structure, thus rendering them inactive.

All the above are general notes and comments.

Consult your Physician to interpret the results and administer the most appropriate treatment regimen for your case.

**IMPORTANT NOTE:** Special laboratory tests are performed for research purposes and as ancillary or supplementary tests in the context of a conventional laboratory test. Special laboratory tests should only be used with other established medical data (e.g., medical history, symptoms, clinical examination, results of different lab tests, etc.).



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Name:	Gender:	Age:
Sample Taken Day:	Date of Analysis:	
Specimen: Whole Blood	Patient Code Number:	

### **Special Comments**

### COPPER (Cu)

Copper is an essential metalloenzyme needed for hemoglobin synthesis. It readily complexes with L-amino acids, which facilitate its absorption from the stomach and duodenum. There are three distinct syndromes of deficiency: The first is characterized by anemia and hypoproteinemia and is easily corrected with combined copper and iron supplementation. The second occurs in malnourished infants, receiving high-calorie, low copper diets. Neutropenia, anemia, diarrhea, bone changes and hypocupremia respond to copper therapy. The third is the genetic defect, Menke's syndrome, in which copper is not absorbed from the intestinal mucosa. Results are low blood, liver, and hair copper levels.

**Deficiency Symptoms**: Reduced hemoglobin synthesis, impaired iron metabolism, hypochromia, microcytic anemia, Kwashiorkor, heart and liver disease, poor growth and development, infertility, pancreatic dysfunction, progressive mental deterioration and defective keratinization of hair.

**Recommended Daily Allowance**: Adults (18 years and older): 900 mcg for adults; 1000 mcg for pregnant women; 1300 mcg for nursing women; 890 mcg for adolescents 14-18 years old. Vegan diets appear to provide adequate amounts of copper.

**Sources**: Liver, shellfish, kidneys, egg yolk, legumes, and nuts.

**Therapeutic Consideration**: Deficiency may be due to a lack of metalloenzymes in the liver. Tyramine (tyrosine + amine) increases copper absorption. Citrus fruits increase the absorption in the small intestine, and glutamine increases copper transport into blood and tissues.

### MANGANESE (Mn)

Manganese is a co-factor for many enzymes including arginase, cholinesterase, phosphoglucomutase, pyruvate carboxylase, mitochondrial superoxide dismutase, and several phosphatases, peptidases and glucosyltransferases. It functions with Vitamin

K in the formation of prothrombin and is needed for the acetylcholine synthesis. Manganese is mostly stored in the liver and the kidneys. Acute deficiency has never been reported in humans, but symptoms of decreased intake include fatigue, lack of physical endurance, hearing loss, slow growth of fingernails and hair, impaired bone metabolism, impaired glucose metabolism

incl. diabetes, reduced fertility, and increased allergic sensitivities. Deficiency symptoms may be caused by dietary insufficiency, intestinal malabsorption, or excess dietary intake of phosphorus, cobalt or magnesium. Manganese is absorbed in the small intestine and excreted in bile and pancreatic secretion.

Sources: Liver, kidney, wheat germ, legumes, black tea and nuts.

Uses/Documentation: Manganese is known to be an important nutrient, but manganese deficiency has not been documented

in humans, as dietary intakes often exceed dietary requirements. The element may be added to TPN solutions in patients who

receive chronic parenteral nutrition.

Contraindications/Precautions:

Note: Manganese supplements should be used cautiously in young children, pregnant and lactating females. Do not supplement manganese in these populations without medical supervision.

### **MOLYBDENUM (Mo)**

ReMolybdenum serves as a co-factor for xanthine and aldehyde oxidases. Dietary molybdenum is readily absorbed by the characters in the urine and bile. Saunders Elsevier

SOURCES: Whole grains, legumes, leafy vegetables and organ meats. The RDA is 0.15 - 0.5 mg/day, depending on age and

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status. Acute deficiency symptoms are unknown in humans. Excess intake of copper, zinc, and sulfates can depress Mo-update, causing disturbances in the uric acid cycle. Low molybdenum levels have been associated with impotency, increased cancer susceptibility, gout, dental caries, defects in the metabolism of sulfur-containing amino acids and asthma tendency.

THERAPEUTIC CONSIDERATION: Increase molybdenum intake and support intestinal function.

#### **SELENIUM (Se)**

Selenium is an essential constituent of the enzyme glutathione peroxidase and is known for its antioxidant properties. It is linked to cysteine as selenocysteine, an enzyme found in blood, liver and other tissues. In humans, toxicity is rare, but excessive intake results in alkali disease, characterized by liver and neuromuscular disorders. Long-term exposure or excess supplementation can cause toxicity symptoms including alopecia, arthritis, atrophic, brittle nails, prevailing garlic breath and body odor, GI disorders, irritability, kidney impairment, metallic taste, and yellowish skin.

Therapeutic Consideration: Sulfates and sulfur-containing amino acids reduce selenium absorption and toxicity. Methionine detoxifies excess selenium.

#### ZINC (Zn)

Zinc is a cofactor for many metalloenzymes, incl. those involving RNA and DNA synthesis. It is necessary for growth, healthy cell division, and insulin production. Pregnant women, cancer and burn patients, are at high risk for zinc deficiency, causing fatigue, poor growth, menstrual problems, and delayed sexual maturity. Deficiency causes are malnutrition and malabsorption. Enteropathic acrodermatitis, skin lesions, diarrhea, anorexia, hair loss, growth retardation, extreme irritability and increased susceptibility to infection are known deficiency symptoms. The zinc absorption occurs mainly in the small intestine, and Vitamin B6 is needed for utilization. Excreted zinc occurs in urine and sweat. The RDA is 3-10 mg/day, depending on age and physiology. A much higher intake is warranted with proper supervision in severe zinc deficiency states. **Source**: Yeast, meat, fish, legumes, and eggs. The zinc in whole grains has a low bio-availability. Phytates block zinc absorption, and a high intake of uncooked grains or unleavened bread can cause zinc deficiency.

**Therapeutic Consideration**: Zinc supplementation with increased vitamin B6 intake. High exposure to toxic metals reduces zinc absorption and increases the need for zinc and vitamin B6 supplementation.

#### MAGNESIUM (Mg)

Magnesium is an essential element with both electrolyte and enzyme-activator functions. It is a predominately intracellular cation, needed for cell function. 1% of body magnesium is found in blood, 60% is stored in bone, and the remainder is equally divided between muscle and other soft tissue. The absorption and excretion of magnesium is regulated by the renal system and parathyroid hormones.

HYPOMAGNESIA and MAGNESIUM DEFICIENCY are rare and generally caused by decreased uptake of magnesium caused by gastrointestinal disorders (steatorrhea, malabsorption syndrome, gut resections, or protein-calorie malnutrition, or increased

urinary losses due to renal disease or high rates of production of aldosterone, hyperparathyroidism and diabetes mellitus. DEFICIENCY SYMPTOMS: Nervous disorders (tixs, tremors, muscle spasms during mild activity), disorientation, cardiac arrhythmia, fast pulse, gastrointestinal problems, pancreatitis, nausea, vomiting, convulsions, seizures (especially in combination with vitamin B6 deficiency)

SOURCE: All plant foods, fish and seafood.

THERAPEUTIC CONSIDERATION: Hypomagnesia is often a symptom of alcoholism, liver cirrhosis, diabetic acidosis, atherosclerosis. Adequate magnesium reduces blood pressure, prevents circulatory problems, headaches, insomnia, excessive perspiration and has an age-retarding effect. Studies indicate that 500-1000 mg/day reduced overall illness in tested

individuals.

#### ANTIMONY (Sb)

Antimony has no known function in living organisms and is not highly toxic. It is found in hair tissue and other organs, with the highest concentration in lymph nodes, lungs, skin and adrenals. Environmental exposure and illness affect the antimony concentration of some tissue. Hair and lung tissue of smelter workers contained high amounts of this trace element and uremic patients have also shown high Sb-levels. Food stored in enamel vessels and cans may contain appreciable antimony concentration. New research indicates that PET (Polyethylenterephthalate) bottles contain appreciable amounts of Sb, and the antimony concentration of mineral water stored in such bottles has been found to increase over time i.e., mineral water

3. Interpretation of Diagnostic Tests, Wallach, 8<sup>th</sup> ed., 2007, Lippincott

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takes up Sb from PET. Trivalent antimony is more toxic than the pentavalent form; however, there is no evidence that this element is carcinogenic.

Therapeutic Consideration: Increase vitamin C and B-complex intake.

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All information is provided exclusively for scientific and educational purposes and does not constitute medical advice or diagnosis

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